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## Boyle's Law

states that as the volume of a gas changes, so does its pressure.


Air inside the syringe is under
$\qquad$ pressure, causing the marshmallow to $\qquad$ .


Air inside the syringe is under ___ pressure, causing the marshmallow to $\qquad$ .

## Charles' Law

states that as the temperature of a gas changes, so does its volume.


If the temperature of a gas then the volume


If the temperature of a gas then the volume

## Boyle's Law

states that as the volume of a gas changes, so does its pressure.


If the volume of a gas (4) then the pressure $(\downarrow$.


Air inside the syringe is under
LOW pressure, causing the marshmallow to EXPAND. .


Air inside the syringe is under HIGH pressure, causing the marshmallow to SHRINK.

## Charles' Law

states that as the temperature of a gas changes, so does its volume.


If the temperature of a gas $(4)$ then the volume $(\mathbb{\top}$.


If the temperature of a gas $\downarrow$, then the volume $\downarrow$.

